

EX. 1:

He : GAZ MONOATOMIQUE \rightarrow G.P.

11

INITIALEMENT: T_A, V_A

T. QUASI-STATIQUE ISOTHERME : $V_A \rightarrow V_B = 2V_A$
[RÉVERSIBLE]

T. " " ADIABATIQUE : $P_B \rightarrow P_C = P_A$
[RÉVERSIBLE]

Q1) $c_v = \frac{3}{2} R \Rightarrow \gamma = \frac{c_p}{c_v} = \frac{c_v + R}{c_v} = 1 + \frac{R}{c_v} = 1 + \frac{R}{\frac{3}{2} R} = \frac{5}{3}$

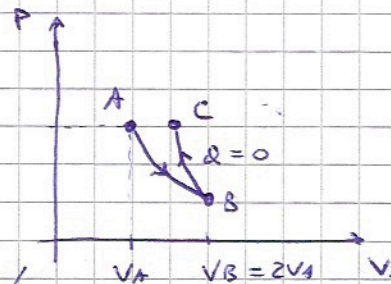
Q2) ISOTH.:

$V_B = 2V_A$

$T_B = T_A$

$P_B = \frac{mRT_B}{V_B} = \frac{mRT_A}{2V_A} = \frac{P_A}{2}$

↑
éq. état



Q3) ADIAB.:

$P_C = P_A$

$PV^\gamma = \text{const} \Rightarrow P_C V_C^\gamma = P_B V_B^\gamma \Rightarrow V_C = V_B \left(\frac{P_B}{P_C} \right)^{\frac{1}{\gamma}} = 2V_A \left(\frac{1}{2} \right)^{3/5}$

$T_C = \frac{P_C V_C}{mR} = \frac{P_A 2V_A (1/2)^{3/5}}{mR} = 2 \left(\frac{1}{2} \right)^{3/5} T_A$

↑
éq. état

Q4) $W = W_{AB} + W_{BC}$

$Q = Q_{AB} + Q_{BC} = Q_{AB}$

0 (ADIAB.)

ISOTH.: $\Delta U_{AB} = m c_v \Delta T_{AB} = 0 \quad (T_B = T_A) \Rightarrow$

$\Rightarrow Q_{AB} = -W_{AB}; W_{AB} = - \int_{V_A}^{V_B} p dV = - m R T_A \int_{V_A}^{V_B} \frac{1}{V} dV =$

$= - m R T_A \ln \left(\frac{V_B}{V_A} \right) = - m R T_A \ln 2$

$$ADIB.: W_{BC} = \Delta U_{BC} = m c_v (T_C - T_B) =$$

$$= m c_v \left[2 T_A \left(\frac{1}{2}\right)^{3/5} - T_A \right] = m c_v T_A \left[2 \left(\frac{1}{2}\right)^{3/5} - 1 \right]$$

$$\Rightarrow W = W_{AB} + W_{BC} = -m R T_A \ln 2 - m c_v T_A \left[1 - 2 \left(\frac{1}{2}\right)^{3/5} \right] =$$

$$= -m R T_A \left\{ \ln 2 + \frac{3}{2} \left[1 - 2 \left(\frac{1}{2}\right)^{3/5} \right] \right\}$$

$c_v = \frac{3}{2} R$

$$Q = Q_{AB} = -W_{AB} = m R T_A \ln 2$$

~~ADIB. process~~

Q5) $V_A = 1 \text{ l}$ $m = 4 \text{ g}$ $R \approx 2077 \frac{\text{J}}{\text{K kg}}$

$T_A = 27^\circ\text{C} = 300 \text{ K}$

$$P_C = P_A = \frac{m R T_A}{V_A} = \frac{4 \cdot 10^{-3} \cdot 2077 \cdot (27 + 273)}{10^{-3}} \text{ Pa} \approx 25 \cdot 10^5 \text{ Pa}$$

$$T_C = 2 \left(\frac{1}{2}\right)^{3/5} T_A \approx 396 \text{ K} \approx 123^\circ\text{C}$$

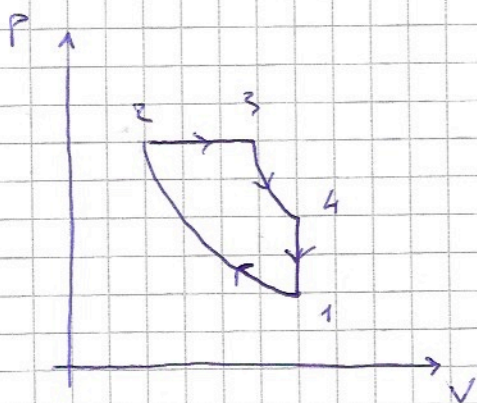
$$V_C = 2 \left(\frac{1}{2}\right)^{3/5} V_A \approx 1,3 \cdot 10^{-3} \text{ m}^3$$

$$W = -m R T_A \left\{ \ln 2 + \frac{3}{2} \left[1 - 2 \left(\frac{1}{2}\right)^{3/5} \right] \right\} \approx -533 \text{ J}$$

$$Q = m R T_A \ln 2 \approx 1728 \text{ J}$$

EX. 2:
Q1)

3



1 → 2: ADIABATIQUE

2 → 3: ISOBARE (COMBUSTION)

3 → 4: ADIABATIQUE

4 → 1: ISOCHORE

CYCLE MOTEUR ⇒ parcouru dans le sens des aiguilles de la montre.

Q2) $Q_{12} = 0$; $Q_{34} = 0$

$Q_{23} = Q_C > 0$ (reçue) ; $Q_{23} = \Delta H_{23}$; H : ENTHALPIE
 $Q_{41} = Q_F < 0$ (cédée) ; $Q_{41} = \Delta U_{41} = m c_V (T_1 - T_4)$ [$V = \text{const}$]

Q3) 1^o PRINCIPE : $\Delta U = W + Q_C + Q_F = 0$ (cycle) ⇒

⇒ $W = -Q_C - Q_F = -(Q_{23} + Q_{41})$
 $W < 0$ (MOTEUR) ⇒ $|Q_{41}| < Q_{23}$

Q4) $\eta = \frac{|W|}{Q_C} = \frac{Q_{23} + Q_{41}}{Q_{23}} = 1 + \frac{Q_{41}}{Q_{23}}$

à partir de Q2) $Q_{23} = \Delta H_{23} = m c_P (T_3 - T_2)$ [$p = \text{const}$]
 $Q_{41} = \Delta U_{41} = m c_V (T_1 - T_4)$ [$V = \text{const}$]

⇒ $\eta = 1 + \frac{c_V (T_1 - T_4)}{c_P (T_3 - T_2)}$

Q5) LOI de LAPLACE : $PV^\gamma = \text{const} \Rightarrow$

$P_i V_i^\gamma = P_f V_f^\gamma \Rightarrow$

⇒ $\frac{P_i}{P_f} = \left(\frac{V_f}{V_i}\right)^\gamma \Rightarrow \frac{T_i}{T_f} = \left(\frac{V_f}{V_i}\right)^{\gamma-1}$

en utilisant l'éq. état pour éliminer P_i, P_f

⇒ $\frac{T_1}{T_2} = \left(\frac{V_2}{V_1}\right)^{\gamma-1}$; $\frac{T_3}{T_4} = \left(\frac{V_4}{V_3}\right)^{\gamma-1} \Rightarrow$

$$\Rightarrow \frac{T_2}{T_1} = \left(\frac{V_2}{V_1} \right)^{\gamma-1} = \left(\frac{1}{\alpha} \right)^{\gamma-1} \quad \alpha = \frac{V_1}{V_2}$$

4

$$\frac{T_3}{T_4} = \left(\frac{V_4}{V_3} \right)^{\gamma-1} = \left(\frac{V_4}{V_2} \frac{V_2}{V_3} \right)^{\gamma-1} = \beta = \frac{V_3}{V_2}$$

$$= \left(\frac{V_1}{V_2} \frac{V_2}{V_3} \right)^{\gamma-1} = \left(\frac{\alpha}{\beta} \right)^{\gamma-1}$$

$V_4 = V_1$ (ISOCHORE)

Q6) $2 \rightarrow 3$: ISOBARIC

$$P_2 = P_3 \Leftrightarrow \frac{mR T_2}{V_2} = \frac{mR T_3}{V_3} \Rightarrow T_3 = T_2 \frac{V_3}{V_2} = T_2 \beta$$

$$Q7) \eta = 1 + \frac{C_V}{C_P} \frac{T_1 - T_4}{T_3 - T_2} = 1 + \frac{1}{\gamma} \frac{T_2 \alpha^{1-\gamma} - T_3 \left(\frac{\alpha}{\beta} \right)^{1-\gamma}}{T_3 - T_2}$$

$$T_1 = T_2 \alpha^{1-\gamma}$$

$$T_4 = T_3 \left(\frac{\alpha}{\beta} \right)^{1-\gamma}$$

$$T_3 = \beta T_2$$

$$= 1 + \frac{1}{\gamma} \frac{T_2 \left[\alpha^{1-\gamma} - \beta \left(\frac{\alpha}{\beta} \right)^{1-\gamma} \right]}{T_2 \left[\beta - 1 \right]} =$$

$$= 1 + \frac{1}{\gamma} \alpha^{1-\gamma} \frac{[1 - \beta^\gamma]}{\beta - 1} \Rightarrow$$

~~scribbles~~

$$\Rightarrow \eta = 1 - \frac{1}{\gamma \alpha^{\gamma-1}} \frac{1 - \beta^\gamma}{1 - \beta}$$

Q8) $\gamma = 1,4$; $\alpha = 1,6$; $\beta = 1,5 \Rightarrow \eta \approx 0,64$